

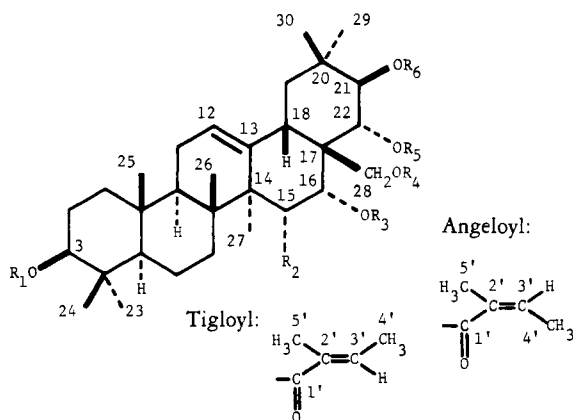
ANTITUMOR AGENTS, 82. ¹ CYTOTOXIC SAPOGENOLS FROM
AESCULUS HIPPOCASTANUM

TAKAO KONOSHIMA and KUO-HSIUNG LEE*

*Natural Products Laboratory, Division of Medicinal Chemistry and Natural Products, School of Pharmacy,
University of North Carolina, Chapel Hill, North Carolina 27514*

ABSTRACT.—Two cytotoxic sapogenols, the new hippocaesculin (**1**) and the known barringtogenol-C 21-angelate (**2**), were isolated from the acid hydrolysis product of a crude saponin fraction that was obtained from the fruits of *Aesculus hippocastanum*. The structures of **1** and **2** were determined from their chemical transformations and spectral data. Compound **1** is either 21-*O*-angeloyl, 22-*O*-tigloyl R₁-barrigenol, or 21-*O*-tigloyl, 22-*O*-angeloyl R₁-barrigenol.

The fruit of *Aesculus hippocastanum* L. (horse chestnuts, Hippocastanaceae) has been used as an herbal remedy for the treatment of mammary indurations and cancer (1). Prior phytochemical studies (2-5) on this plant have yielded a saponin, aescin, which is a mixture of the acylated glycosides of protoaescigenin and barringtogenol-C, with angelic, tiglic, and acetic acids as the acyl groups. As a result of our continuing searches among medicinal plants for novel, naturally occurring, potential antitumor agents, the acid hydrolyzed product of an *n*-BuOH extract of the fruits of *A. hippocastanum* was found to show significant in vitro cytotoxicity in the 9-KB (human nasopharyngeal carcinoma) cell culture assay (6). Bioassay-directed fractionation of the foregoing active extract led to the isolation and characterization of two cytotoxic sapogenols, the new hip-



	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆
1	H	OH	H	H	angeloyl*	tigloyl*
2	H	H	H	H	H	angeloyl
3	H	OH	H	H	H	H (R ₁ -barrigenol)
4	H	H	H	H	H	H (barringtogenol-C)
5	Ac	OAc	H	Ac	angeloyl*	tigloyl*
6	Ac	OAc	H	Ac	Ac	Ac
7	H	-O(CH ₃) ₂ C-		-(CH ₃) ₂ C-		H
8	Ac	-O(CH ₃) ₂ C-		-(CH ₃) ₂ C-		Ac
9	H	OH	H	H	angeloyl	angeloyl
10	Ac	H	H	Ac	Ac	angeloyl
11	Ac	H	H	Ac	Ac	Ac
12	H	H	H	-(CH ₃) ₂ C-		H

*In these compounds, the angeloyl and the tigloyl moieties are interchangeable.

¹For part 81, see N. Fukamiya and K.H. Lee, *J. Nat. Prod.*, **49**, 348 (1986).

pocaesculin [**1**, ED_{50} (KB)=3.6 μ g/ml] and the known barringtonenol-C 21-angelate [**2**, ED_{50} (KB)=3.0 μ g/ml]. Compound **2** was previously isolated from the leaves of *Pittosporum tobira* (7); however, its cytotoxicity is revealed for the first time.

On alkaline hydrolysis, **1** and **2** gave the known sapogenins R₁-barrigenol (**3**) (7) and barringtonenol-C (**4**) (7), respectively. Compounds **3** and **4** were inactive in the KB test system ($ED_{50} \geq \mu$ g/ml), but the acetonide derivative (**7**) of **3** showed significant cytotoxicity with an ED_{50} =4.0 μ g/ml.

RESULTS AND DISCUSSION

On acid hydrolysis, the crude saponin fraction that was obtained from a methanolic extract of fruits of *A. hippocastanum* afforded a mixture of sapogenols. The mixture was chromatographed repeatedly on a silica gel column and purified by preparative tlc to give **1** and **2**.

Compound **1**, C₄₀H₆₂O₈, showed ir absorption bands (1690, 1640, and 1260 cm⁻¹) due to an α , β -unsaturated ester. Compound **1** gave a triacetate (**5**) upon acetylation.² Upon hydrolysis with 3% KOH in MeOH, **1** afforded a triterpene **3**. The ¹³C-nmr spectrum of **3** (Table 1) showed the presence of five secondary alcohols at δ 78.1, 67.5, 72.4, 78.4, 77.2, and one primary alcohol at δ 67.8. Compound **3** was converted to a pentaacetate (**6**), a diacetonide (**7**), and a diacetonide diacetate (**8**) by usual methods. Compounds **3**, **7**, and **8** exhibited prominent mass spectral peaks at m/z 298, 378, and 420, respectively, due to a retro Diels-Alder cleavage of ring-C in each compound. The characteristic pairs of fragments corresponding to the release of acetonide moieties from rings D and E of **7** and **8** were also observed at m/z 320 and 262 as well as

TABLE 1. ¹³C-nmr Spectra^a of **1**, **2**, **3**, and **4**

Carbon No.	Compounds				Carbon No.	Compounds			
	1	3	2	4		1	3	2	4
1	39.3 ^b	39.3	39.2	39.1	21	79.0	78.4	82.0	78.6
2	28.2	28.2	28.1	28.1	22	73.6	77.2	73.3	77.2
3	78.1	78.1	78.1	78.0	23	28.7	28.8	28.7	28.7
4	39.3 ^b	39.4	39.4	39.3	24	16.6	16.6	16.6	16.5
5	55.6	55.7	55.8	55.8	25	15.9	16.0	15.8	15.8
6	19.1	19.2	18.8	18.8	26	17.6	17.7	17.0	17.0
7	36.8	36.8	33.3	33.2	27	21.1	21.1	27.4	27.4
8	41.5	41.5	40.1	40.1	28	63.1	67.8	66.2	68.4
9	47.3	47.5	47.2	47.1	29	29.5	30.6	29.8	30.5
10	37.4	37.4	37.5	37.2	30	20.1	19.4	20.4	19.4
11	24.0	24.1	23.9	23.9	1'	168.1		168.6	
12	125.5	124.5	123.1	122.9	2'	129.4		129.5	
13	143.6	144.7	143.5	143.8	3'	136.7		135.9	
14	47.8	47.5	41.9	42.0	4'	15.6		15.9	
15	67.6	67.5	34.4	34.2	5'	21.1		21.0	
16	73.3	72.4	67.9	67.8	1''	167.9			
17	48.4	48.2	48.2	47.2	2''	129.1			
18	41.0	42.1	40.5	41.2	3''	137.0			
19	46.9	47.9	47.9	48.2	4''	12.4			
20	36.5	36.4	36.1	36.3	5''	13.9			

^aRecorded in pyridine-*d*₅. Chemical shifts are in δ (ppm). The signal assignments were referred to DEPT experiments.

^bThese signals were overlapping.

²The lack of acetylation of the OH at C-16 is due to its steric hindrance with Me-14, H α -19, and H α -21.

362 and 304, respectively. The foregoing evidence suggested that one of the six hydroxyl groups was located at either ring A or B. The remainder were at rings D and E, and among them, four form two pairs of α -glycols. The presence of two pairs of doublets at δ 4.39 (H_{β} -15) and 4.93 (H_{β} -16) with $J=6.7$ Hz, and δ 3.69 (H_{β} -22) and 4.12 (H_{β} -21) with $J=10.1$ Hz (Table 2) in the 1H -nmr spectrum of **7** confirmed this latter assignment, and led to the placement of these five hydroxyl groups at C-15, C-16, C-21, C-22, and C-28. A comparison of the 1H -nmr, ^{13}C -nmr, and ms data between **3** and R_1 -barrigenol (8-10) established their identity.

The presence of one tigloyl and one angeloyl ester group in **1** was suggested by the characteristic 1H -nmr signals for the acid portion of these esters at δ 6.80 (1H, m, tigloyl H-3') and 1.76 (3H, d, $J=6.7$ Hz, tigloyl Me-3'), as well as δ 6.09 (1H, m, angeloyl H-3') and 1.92 (3H, d, $J=7.0$ Hz, angeloyl Me-3'). This was further supported by the prominent mass peaks in **1** at m/z 462, 362, and 262, which were due to the D/E ring residue and its subsequent release of the acyl moiety as shown in Table 3.

That the two acyl groups were located at C-21 and C-22 was based upon the fact that in the 1H -nmr spectrum of the triacetate **5**, H-21 and H-22 were shown as two pairs of doublets at δ 5.87 and 5.54, respectively, whereas in the R_1 -barrigenol pentaacetate **6**, the two pairs of doublets were shifted upfield to δ 5.66 (H-21) and 5.37 (H-22). The other protons attached to carbons bearing acetyl groups at C-15, C-16, and C-28 were seen at δ 5.10, 4.25, and 3.77 and 3.93, respectively, in **5** and at δ 5.09, 4.18, and 3.72 and 3.92, respectively, in **6** and were comparable in their chemical shifts (Table 2).

Since the ^{13}C -nmr spectrum of **1** (hippocaesculin) was superimposable with that of 21, 22-di-*O*-angeloyl R_1 -barrigenol (**9**) obtained from the fruits of *Xanthoceras sorbifolia* (11), except for the acyl moieties (Table 1), this led to the confirmation of the structure of hippocaesculin as depicted in **1**, in which the angeloyl and the tigloyl groups located at the C-22 and C-21 positions are interchangeable.³

Compound **2** possessed four secondary alcoholic groups (δ 78.1, 67.9, 82.0, and 73.3) and one primary alcohol (δ 66.2) as shown in its ^{13}C -nmr spectrum and gave a triacetate **10** upon acetylation. Alkaline hydrolysis of **2** yielded the known sapogenin **4**(7), which in turn was converted to a tetraacetate (**11**) and a monoacetone (**12**) by standard methods. A comparison of the 1H -nmr spectra of **11** and **12** with those reported in the literature (12, 13) led to the characterization of **4** as barringtonol-C.

The presence of an angeloyl group in **2** was revealed by an ir band at 1680 (α,β -unsaturated ester) cm^{-1} ; five carbon signals at δ 168.6 (C-1'), 129.5 (C-2'), 135.9 (C-3'), 15.9 (C-4'), and 21.0 (C-5') in the ^{13}C -nmr spectrum (Table 1); and by a diagnostic mass peak at m/z 264 [(a)-angeloyl as shown in Table 3] in the mass spectrum. A comparison of the ^{13}C -nmr spectra between **2** and **4** showed that the signals of C-21 and C-22 in **2** were shifted downfield by 3.4 ppm and upfield by 3.9 ppm, respectively, compared to those in **4**, due to an acylation shift (Table 1). Thus, compound **2** was characterized as 21-*O*-angeloyl barringtonol-C based upon evidence described above.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Melting points were determined on a Fischer-Johns melting point apparatus and are uncorrected. Ir spectra were measured on Perkin-Elmer 1320 ir spectrometer. 1H -nmr and ^{13}C -nmr spectra were recorded on a Bruker 250 MHz spectrometer using TMS as an internal standard. Mass spectra were determined on a V. G. Micromass 70-70 instrument at 70 eV using a direct inlet system. Silica gel (Merck, type G 60, 70-230 mesh) was used for column chromatography, and precoated silica gel plates were used for analytical (Merck, 60 F-254, 0.25 mm) and preparative (Analtech, G, 1000 m) tlc. Compounds were visualized by uv light or spraying with 15% H_2SO_4 solution followed by heating.

³Structural determination of **1** by single crystal X-ray crystallographic analysis is in progress.

TABLE 2. Nmr Spectral Data of **5**, **6**, **7**, **8**, **10**, **11**, and **12**^a

Compounds	H _α -3	H-12	H _β -15	H _β -16	H _α -21 ^b	H _β -22 ^b	H _γ -28	Acyl moieties
5	4.50 t-like	5.49 t-like	5.10 d, J=3.7	4.25 brs	5.87 d, J=10.3	5.54 d, J=10.3	3.77, 3.93 ABd, J=11.7	6.80(1H, m), 5.98(1H, dq, J=7.0, 1.5) 1.74(3H, d, J=7.0), 1.91(3H, d, J=7.0) 1.80(6H, s), 2.04, 2.07, 2.10 (each 3H, s, COCH ₃ × 3)
6	4.50 t-like	5.50 t-like	5.09 d, J=3.9	4.18 brs	5.66 d, J=10.2	5.37 d, J=10.2	3.71, 3.92 ABd, J=11.7	2.02(6H, s), 2.05, 2.07, 2.09 (each 3H, s, COCH ₃ × 5)
7	3.20 t-like	5.32 t, J=3.4	4.39 d, J=6.7	4.93 d, J=6.7	4.12 d, J=10.1	3.69 d, J=10.1	3.41, 3.48 ABd, J=11.8	
8	4.48 t-like	5.32 t-like	4.38 d, J=7.0	4.93 d, J=7.0	5.59 d, J=10.5	3.82 d, J=10.5	3.40, 3.49 ABd, J=11.8	2.04, 2.05 (each 3H, s, COCH ₃ × 2)
10	4.50 t, J=8.0	5.37 t-like		4.20 brs	5.66 d, J=10.2	5.45 d, J=10.2	3.68 (2H) brs	
11	4.50 t, J=8.0	5.36 t-like		4.18 brs	5.53 d, J=10.2	5.41 d, J=10.2	3.66 (2H) brs	6.05(1H, dq, J=9.0, 1.5), 1.94(3H, d, J=9.0), 1.86(3H, br s), 2.05(9H, s, COCH ₃ × 3)
12	3.21 t-like	5.29 t-like		4.78 brs	4.15 d, J=10.2	3.73 d, J=10.2	3.38, 3.58 Bd, J=11.7	2.01, 2.02, 2.05, 2.06 (each 3H, s, COCH ₃ × 4)

^aChemical shifts are shown in δ-values (ppm) with coupling constants (J) in Hz. The abbreviations brs, d, t, dq, and m refer to broad singlet, doublet, triplet, doublet of quartet, and multiplet, respectively.

^bH_α-21 is shifted downfield by the adjacent O_α-16 compared to H_β-22.

TABLE 3. Mass Spectral Fragmentation of 1, 2, 3, and 4

Compounds		M ⁺	M ⁺ -H ₂ O		M ⁺ -acyl	(a)-H ₂ O		(b)-H ₂ O		base peak
R ₁	R ₂		R ₃	M ⁺ -2H ₂ O		(a)	(a)-acyl	(b)	(b)-acyl	
1	OH	Angeloyl ^a tigloyl	H	670 ^b	570 470	462	444 362	344 244	207 189	83 43
2	H	Angeloyl	H	572	472	364	346 264	246	207 189	83
4	H	H	H	490	472 454	282	264 246	246	207 189	215

^aAngeloyl acid or tiglic acid: CH₃CH=C(CH₃)COOH *m/z* 100 and CH₃CH=C(CH₃)CO- *m/z* 83.^b*m/z*.

PLANT MATERIAL.—The fruits of *A. hippocastanum* were purchased in January 1985, from Wilcox Drug Company, Inc., Boone, North Carolina. A voucher specimen was placed on file at Wilcox Drug Co.

EXTRACTION OF CRUDE SAPONIN FRACTION.—The crushed fruit of *A. hippocastanum* (7.5 kg) was defatted with *n*-hexane (5 liters×2) at room temperature and extracted with hot MeOH (5 liters×3). The MeOH solution was concentrated in vacuo to yield a residue that was then suspended in H₂O (1.2 liters). The suspended aqueous layer was washed with Et₂O and EtOAc, and extracted with *n*-BuOH. Evaporation of the *n*-BuOH extract under reduced pressure gave the crude saponin fraction as a yellowish gum (230 g).

ACID HYDROLYSIS OF THE CRUDE SAPONIN FRACTION.—A solution of *n*-BuOH extract (6 g) in 1 N HCl-EtOH (1:1, 110 ml) was refluxed for 2 h. The reaction mixture was diluted with 30 ml of H₂O and concentrated to the half volume in vacuo to afford a brown precipitate. The precipitate was washed with H₂O and dried to yield white powder (2 g). Repeated column chromatography of this white powder on silica gel [MeOH-CHCl₃ (5:95)], followed by recrystallization of the resulting major fraction from MeOH/H₂O furnished colorless needles of **1** (23 mg, 0.00031% yield) and **2** (35 mg, 0.00047% yield).

Compound 1.—Mp 254-256°; [α]_D+25° (c, 0.53, MeOH); ir (KBr) cm⁻¹ 3400 (OH), 1690, 1640, and 1260; ¹H-nmr δ0.79, 0.91, 0.94 (3H each, s, Me×3), 0.99 (6H, Me×2), 1.08, 1.38 (3H each, s, Me×2), 1.76 (3H, d, *J*=6.7 Hz, tigloyl Me-3'), 1.92 (3H, d, *J*=7.0 Hz, angeloyl Me-3'), 1.80 (6H, br s, tigloyl, and angeloyl Me-2'×2), 6.09 (1H, m, angeloyl H-3'), and 6.80 (1H, m, tigloyl H-3'); ¹³C-nmr (see Table 1); ms *m/z* 670 (M⁺, weak), 652.4337 [calcd for C₄₀H₆₀O₇ (M⁺-H₂O): 652.4369]; other peaks, see Table 3.

Compound 2.—Mp 242-244°; [α]_D+31° (c, 0.62, MeOH); lit (7) reported mp 252-254° and [α]_D+31.7° (MeOH) for 21-*O*-angeloylbarringtonenol-C; ir (KBr) cm⁻¹ 3400 (OH), 1680, 1260; ¹H-nmr δ0.79, 0.90, 0.94, 1.02, 1.01, 1.43 (21 H, s each, Me×7), 1.95 (3H, br s, angeloyl Me-2'), 2.01 (3H, d, *J*=7.1 Hz, angeloyl Me-3'), and 6.09 (1H, dq, *J*=7.1 and 1.5 Hz, angeloyl H-3'); ¹³C-nmr (see Table 1); ms, see Table 3.

ISOLATION OF COMPOUNDS **3** AND **4**.—The white powder (500 mg) obtained by acid hydrolysis of the crude saponin fraction was hydrolyzed with 3% KOH in MeOH (50 ml) under reflux for 1 h. The reaction mixture was concentrated and extracted with EtOAc. The organic layer was washed with H₂O and dried over anhydrous MgSO₄. The solvent was evaporated in vacuo to dryness to afford 200 mg of yellow powder. This powder was chromatographed on silica gel [CHCl₃-MeOH-H₂O (9:1:0.1), lower layer] to afford two major fractions. These fractions were purified by recrystallization from MeOH/H₂O to afford compound **3** (R₁-barrigenol) (10) and compound **4** (barringtonenol-C) (13) as colorless needles.

Compound 3.—Mp >300°; [α]_D+38° (c, 0.54, MeOH); lit (7) reported mp 310-312° and [α]_D+40.7° (dioxane); ir (KBr) cm⁻¹ 3400 (OH); ¹H-nmr (pyridine-*d*₅)⁴ δ0.99, 1.07, 1.12, 1.24, 1.36, 1.41, 1.88 (3H each, s, Me×7), 3.49 (1H, br s, H_α-3), 3.79, 4.13 (2H, ABq, *J*=10.4 Hz, H₂-28), 4.45 (1H, d, *J*=4.5 Hz, H_β-15), 4.63 (1H, d, *J*=9.6 Hz, H_β-22), 4.84 (1H, d, *J*=9.6 Hz, H_α-21), 4.98 (1H, d, *J*=4.5 Hz, H_β-16), and 5.54 (1H, t-like, H-12); ¹³C-nmr (see Table 1); ms *m/z* 506.3621 [calcd for C₃₀H₅₀O₆ (M⁺): 506.3605]; other peaks, see Table 3.

Compound 4.—Mp 275-280°; [α]_D+27° (c, 0.51, MeOH); ir (KBr) cm⁻¹ 3400 (OH); ¹H-nmr (pyridine-*d*₅) δ0.97, 0.98, 1.06, 1.24, 1.34, 1.40, 1.86 (3H each, s, Me×7), 3.49 (1H, br s, H_α-3), 3.75, 4.05 (2H, ABd, *J*=10.0 Hz, H₂-28), 4.64 (1H, d, *J*=9.5 Hz, H_β-22), 4.80 (1H, d, *J*=9.5 Hz, H_α-21), and 5.45 (1H, t-like, H-12); ¹³C-nmr see Table 1; ms *m/z* 490.3654 [calcd for C₃₀H₅₀O₅ (M⁺): 490.3655]; other peaks, see Table 3.

GENERAL PROCEDURE FOR ACETYLATION.—The acetates described below were prepared by acetylation of compounds **1-4** and **7** in Ac₂O in pyridine at room temperature for 12-18 h followed by usual work up and purification by either preparative tlc or column chromatography to yield the pure compounds.

Triacetate 5.—Compound **5** (34 mg) was prepared from **1** (60 mg) as colorless needles: mp 263-265°; ir (KBr) cm⁻¹ 1730 (acetyl CO), 1690, 1650 (α,β-unsaturated ester), 1250 (RCOOC); ¹H-nmr, see Table 2. *Anal.* Found: C 69.64; H, 8.62; calcd for C₄₆H₆₈O₁₁: C, 69.32; H, 8.60%.

Pentaacetate (6, R₁-barrigenol pentaacetate).—Compound **6** (20 mg) was prepared from **3** (54 mg) as white amorphous powder: ir (CHCl₃) cm⁻¹ 3660 (OH, weak), 1740 (acetyl CO), 1260 (RCOOC); ¹H-nmr, see Table 2, (10). *Anal.* Found: C, 66.77; H, 8.32; calcd for C₄₀H₆₀O₁₁: C, 67.01%; H, 8.44%.

⁴Lit (10) reported ¹H-nmr (pyridine-*d*₅, 100 MHz) spectrum of R₁-barrigenol at δ0.99, 1.06, 1.12, 1.24, 1.35, 1.39, 1.86 (3H, each s), 3.44 (1H, t-like, H_α-3), 3.80, 4.12 (2H, ABq, *J*=11.0 Hz, H₂-28), 4.44 (1H, d, *J*=4.0 Hz, H_β-15), 4.96 (1H, d, *J*=4.0 Hz, H_β-16), 4.61 (1H, d, *J*=10.0 Hz, H_β-22), and 4.81 (1H, d, *J*=10.0 Hz, H_α-21).

Triacetate 10.—Acetylation of **2** (60 mg) yielded **10** (24 mg) as colorless needles: mp 276–278°; ir (KBr) cm^{-1} 1740, 1720 (acetyl CO), 1690 (α, β -unsaturated ester), 1240 (RCOOC); $^1\text{H-nmr}$, see Table 2. *Anal.* Found: C, 70.15%; H, 8.69%; calcd for $\text{C}_{41}\text{H}_{62}\text{O}_9$: C, 70.45; H, 8.94%.

Tetraacetate (11, *barringtogenol tetraacetate*) (12, 13).—Acetylation of **4** (45 mg) gave **11** (22 mg) as colorless needles: mp 223–225°; ir (CHCl_3) cm^{-1} 3660 (OH, weak), 1730 (acetyl CO), 1250 (RCOOC); $^1\text{H-nmr}$, see Table 1). *Anal.* Found: C, 69.53; H, 8.68; calcd for $\text{C}_{38}\text{H}_{58}\text{P}_9$: C, 69.27; H, 8.87%.

Diacetonide diacetate (8) (7).—Compound **8** (25 mg) was obtained from the diacetonide (**7**, 50 mg) as white amorphous powder: ir (CHCl_3) cm^{-1} 1730 (acetyl CO), 1250 (RCOOC), 1030 (C–O–C); $^1\text{H-nmr}$, see Table 2; ms m/z 655 (M^+ -Me), 612 (M^+ - Me_2CO), 554 (M^+ - $2 \times \text{Me}_2\text{CO}$), 610 (M^+ -HOAc), 420 (DE ring residue), 362 (420- Me_2CO), 304 (420- $2 \times \text{Me}_2\text{CO}$), 250 (AB ring residue), 190 (AB-HOAc).

Acetonide of 3.—To a solution of **3** (100 mg) in anhydrous Me_2CO (50 ml) was added 2,2-dimethoxypropene (1 ml) and *p*-toluenesulfonic acid (15 mg). The mixture was stirred at room temperature for 2 h; then 3 drops of pyridine was added and evaporated in vacuo to yield a residue. The residue was dissolved in a trace of MeOH and poured into ice H_2O . The resulting precipitate was washed with H_2O , purified by column chromatography on neutral alumina, and recrystallized from Me_2CO to afford the diacetonide (**7**, 55 mg) as colorless prisms: ir (CHCl_3) cm^{-1} 3600 (OH, weak) and 1030 (C–O–C), $^1\text{H-nmr}$, see Table 2; ms m/z 586 (M^+ , weak), 571 (M^+ -Me), 528 (M^+ - Me_2CO), 470 (M^+ - $2 \times \text{Me}_2\text{CO}$), 378 (DE ring residue), 320 (378- Me_2CO), 262 (378- $2 \times \text{Me}_2\text{CO}$), 207 (AB ring residue).

Acetonide of 4.—Treatment of **4** (100 mg) in an analogous manner to that described above for the preparation of **7**, afforded a monoacetonide (**12**, 40 mg) as amorphous powder: ir (CHCl_3) cm^{-1} 3500 (OH, weak) and 1030 (C–O–C); $^1\text{H-nmr}$, see Table 2. *Anal.* Found: C, 74.56; H, 10.35; calcd for $\text{C}_{33}\text{H}_{54}\text{O}_5$: C, 74.67; H, 10.26%.

ACKNOWLEDGMENTS

This investigation was supported by a grant from the National Cancer Institute (CA 17625) awarded to K.H. Lee. The authors thank Dr. Y.C. Cheng and Mr. Michael Fisher of the Cancer Research Center, Dr. David L. Harris of the Department of Chemistry and Dr. Michael T. Harvey of the School of Public Health, University of North Carolina at Chapel Hill, for biological assay, nmr, and ms, respectively.

LITERATURE CITED

1. J.L. Hartwell, "Plants Used Against Cancer," Quarterman Publications, Inc., Lawrence, MA, 1982, p. 250, and literature cited therein.
2. G. Wulff and R. Tschesche, *Tetrahedron*, **24**, 415 (1969).
3. R. Kuhn and I. Low, *Tetrahedron*, **22**, 1899 (1966).
4. G. Cainelli, A. Melera, D. Arigoni, and O. Jeger, *Helv. Chim. Acta*, **40**, 2390 (1957).
5. I. Yoshioka, A. Matsuda, K. Imai, T. Nishimura, and I. Kitagawa, *Chem. Pharm. Bull.*, **19**, 1200 (1971).
6. R.I. Geran, N.H. Greenberg, M.M. MacDonald, A.M. Shumacher, and B.J. Abbott, *Cancer Chemother. Rep.*, **3**, 1 (1972).
7. I. Yoshioka, I. Kitagawa, K. Hino, A. Matsuda, and Y. Nakagawa, *Chem. Pharm. Bull.*, **16**, 190 (1968).
8. I. Yoshioka, K. Hino, A. Matsuda, and I. Kitagawa, *Chem. Pharm. Bull.*, **20**, 1499 (1972).
9. S. Ito, T. Ogino, H. Sugiyama, and M. Kodama, *Tetrahedron Lett.*, 2289 (1967).
10. Y. Chen, T. Takeda, Y. Ogiwara, and Y. Iitaka, *Chem. Pharm. Bull.*, **32**, 3378 (1984).
11. Y. Chen, T. Takeda, and Y. Ogiwara, *Chem. Pharm. Bull.*, **33**, 127 (1985).
12. I. Yoshioka, T. Nishimura, A. Matsuda, and I. Kitagawa, *Chem. Pharm. Bull.*, **18**, 1610 (1970).
13. I. Yoshioka, T. Nishimura, A. Matsuda, and I. Kitagawa, *Chem. Pharm. Bull.*, **18**, 1621 (1970).

Received 10 February 1986